1. Photon Energy.

$$E \propto f$$
 gives

$$h = plank \ constant$$

$$E = hf = \frac{hc}{\lambda}$$

$$> 1eV = 1.6 \times 10^{-19} J$$

2. Photo Electric Equation.

$$photon E = workfunction + K.e$$

$$hf = \varphi + \frac{1}{2}m \ v_{max}^2$$

3. Threshold Frequency.

$$f_o = \frac{\varphi}{h}$$

4. Energy levels and photon energy.

$$hf = E_1 - E_2$$

5. De Broglie's Wavelength.

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

6. Electron Diffraction.

$$\lambda = 2d \sin \theta$$

7. Speed of Electron.

$$eV = \frac{1}{2}mv^2$$

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D. H. Whiffen

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